Infrared study of relevance of Nd₂CuO₄ structure to superconductivity of La-Sr-Cu-O system

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A comparative study of the infrared spectra of $Nd_2Cu_{1-x}Ag_xO_4$ and $La_{1.8}Sr_{0.2}Cu_{1-x}Ag_xO_4$ is reported. It is shown that the appearance of an absorption peak at ~680 cm⁻¹ is due to local distortions around the Ag impurities, and the appearance of this peak need not imply a structural transformation from Nd_2CuO_4 type of K_2NiF_4 type. It is shown that Nd_2CuO_4 structure is not important for the superconductivity of La-Sr-Cu-O system. © 1999 Kluwer Academic Publishers

1. Introduction

After the exciting discovery of superconductivity in La_{1.8}Ba_{0.2}CuO₄ by Bednorz and Muller [1], there has been great interest in oxide superconductors both from applied as well as basic points of view. Although there are now several different superconducting oxide systems known, an understanding is still lacking due to conflicting experimental evidences. Infrared reflectance and transmittance experiments [2-4] have shown energy gaps in the range $2\Delta/k_BT_c \sim 1.3-2.7$ well below the BCS prediction of $2\Delta/k_BT_c = 3.53$. Tunneling measurements [5–8] have shown gaps in the range $2\Delta/k_BT_c \sim 3.5-6$, at least as large as the BCS prediction. No isotope shift has been observed [9] in YBa₂Cu₃O₇, while very small shift has been observed in La-Sr-Cu-O system [10]. These facts suggest that superconductivity of these systems may not be of the conventional BCS type. The superconductivity of Y-Ba-Cu-O and La-Sr-Cu-O systems has been related to the antiferromagnetism of the parent compounds [11–13]. We may note, however, that all these systems contain planes of Cu-O to which the superconductivity seems to be connected. In the present context, it becomes important to investigate the oxide superconductors using various techniques that can probe the electronic system and the lattice and their interaction.

Shortly after the discovery of La-Sr-Cu-O system, Stavola *et al.* [14] reported an interesting result from infrared absorption measurements on $La_{2-x}Sr_xCuO_4$. Of the two strong absorption peaks at $v_1 = 677$ cm⁻¹ and $v_2 = 517$ cm⁻¹, the first one gradually decreases in intensity and finally disappears at x = 0.15 which is found to be the optimum concentration for the occurrence of superconductivity. Ohbayashi *et al.* [15] have compared the IR spectra of $La_{1.8}Ba_{0.2}CuO_4$ to that of tetragonal (T_t phase) La_2CuO_4 and tetragonal Nd₂CuO₄ (T_x phase) and they conclude that $La_{1.8}Ba_{0.2}CuO_4$ has Nd₂CuO₄ type structure and not K₂NiF₄ type. They go further to say that the Nd₂CuO₄ structure of La_{1.8}Ba_{0.2}CuO₄ is important for the superconductivity of this compound. It is natural that their arguments must also apply to La_{1.8}Ba_{0.2}CuO₄ as well, since the IR spectra are essentially similar. In this paper, through a comparative study of Ag substituted La_{1.8}Ba_{0.2}CuO₄ and Nd₂CuO₄, we show that the above conclusion of Ohbayashi *et al.* is not valid. Our data shows that La_{1.8}Ba_{0.2}CuO₄ does not necessarily have Nd₂CuO₄ structure, and even if it does, the structure is not important for superconductivity of the La-Sr-Cu-O system. The features of the IR spectra may be explained on the basis of local distortions produced around the Ag impurities.

2. Experimental

The samples of Nd₂Cu_{1-x}Ag_xO₄ (x = 0.05, 0.1) were prepared by mixing Nd₂O₃, CuO and Ag₂O in stoichiometric proportions, calcining them at 700 °C for 6 h and then at 1000 °C for 12 h. Afterwards they were pressed into pellets and heated for 4 h at 1050 °C. From X-ray diffraction the samples were found to be single phase with Nd₂CuO₄ structure. The lattice parameters are found to be a = 3.943 Å and c = 12.154 Å for x = 0.05 and a = 3.945 Å and c = 12.157 Å for x = 0.1samples. Infrared spectra were recorded in KBr matrix using a commercial cw IR spectrometer. The samples for IR were prepared by mixing 0.1 wt % of the sample with KBr and grinding and pressing into pellets.

3. Results and discussion

The IR study of $La_{1-x}Sr_xCuO_4$ system by Stavola *et al.* reveals that the v_1 line of La_2CuO_4 gradually decreases in intensity and finally disappears at x = 0.15. This concentration also happens to be the optimum for the occurance of superconductivity. Fig. 1 shows the IR spectra of $La_{1.8}Sr_{0.2}Cu_{1-x}Ag_xO_4$ compounds. We find that



Figure 1 Infrared spectra of $La_{1.8}Sr_{0.2}Cu_{1-x}Ag_xO_4$ compounds.



Figure 2 Electrical resistivity of $La_{1.8}Sr_{0.2}Cu_{1-x}Ag_xO_4$ as a function of temperature. Note that T_c is nearly the same for x = 0.0 and 0.05 while for x = 0.1 it is clearly lower. The transition becomes broader with increasing Ag concentration.

the v_1 line reappears upon Ag substitution. This has been attributed by us to local distortions around the Ag ions [16]. These defects give rise to local charge disproportionation and hence to some infrared activity. The Ag impurities lower the T_c slightly and make the transition broader (see Fig. 2). Ohbayashi *et al.* have presented in their paper [15] the IR spectra of La₂CuO₄ which changes from an orthorhombic phase to a tetragonal phase (T_t) that is of K₂NiF₄ structure. The structure of La_{1.8}Ba_{0.2}CuO₄ has been deduced from X-ray diffraction to be K₂NiF₄(T_t) type. However, Ohbayashi et al. propose another tetragonal structure (T_x phase) for La_{1.8}Ba_{0.2}CuO₄ which cannot be easily distinguished from T_t phase by X-ray diffraction. According to Ohbayashi et al. the striking difference between T_t and T_x phases is that the T_t phase shows two absorption peaks (at $v_1 \sim 680 \text{ cm}^{-1}$ and $v_2 \sim 510 \text{ cm}^{-1}$) while in the T_x phase the v_1 line is absent. The T_x phase, they point out, has the same structure as Nd₂CuO₄ as this compound also has a similar IR spectrum. From this they go further to say that the superconductivity is specifically related to the Nd₂CuO₄ type structure of La_{1.8}Ba_{0.2}CuO₄. Naturally the same arguments also apply to the La-Sr-Cu-O system. If we apply the arguments of Ohbayashi et al. to $La_{1.8}Sr_{0.2}Cu_{1-x}Ag_xO_4$ system it implies that by Ag substitution, we have switched the structure from T_x to T_t and yet retained superconductivity in the system. This immediately contradicts their contention that Nd₂CuO₄ structure is important for the superconductivity. We therefore, suggest that superconductivity is not related to the Nd₂CuO₄ structure at all. What, however, is related to superconductivity is the drastic change in the electronic structure brought about by Sr substitution which turns the material into a metallic one. Therefore, it appears that it is incorrect to correlate the appearance/disappearance of the v_1 line to a structural transformation. The explanation based on local distortions is more likely from this point of view. The results of IR absorption on $Nd_2Cu_{1-x}Ag_xO_4$ strongly support this explanation. Nd₂CuO₄ shows an absorption peak at around 510 cm⁻¹ and no line corresponding to v_1 [17]. We find that upon Ag substitution a second absorption line appears (see Fig. 3) which corresponds to the v_1 line. From the data we find that $v_1 = 675 \text{ cm}^{-1}$ for x = 0.05 and 670 cm⁻¹ for x = 0.1 compounds. Since we are directly substituting Ag into the T_x phase, we



Figure 3 Infrared spectra of $Nd_2Cu_{1-x}Ag_xO_4$ compounds.

can assume that the Ag substituted compounds are of the same structure as the parent Nd₂CuO₄. The fact that a v_1 line appears in the T_x phase implies that the absence of this line is not characteristic of the T_x phase. Rather, it appears due to local distortions of the lattice produced by the Ag impurities.

4. Conclusions

To conclude, we have shown that in Nd₂CuO₄ an absorption line at about 680 cm⁻¹ appears upon Ag substitution due to local distortions around the Ag impurities. By relating this data to that on superconducting $La_{1.8}Sr_{0.2}Cu_{1-x}Ag_xO_4$ we have shown that Nd₂CuO₄ structure is not essential for the superconductivity of La-Sr-Cu-O and La-Ba-Cu-O systems as suggested by Ohbayashi *et al.* [15].

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References

- 1. J. G. BEDNORZ and K. A. MULLER, Z. Phys. B64 (1986) 189.
- D. A. BONN, J. E. GREEDAN, C. V. STAGER, T. TIMUSK, M. G. DOSS, S. L. HERR, K. KAMARAS, C. D. PORTER, D. B. TANNER, J. M. TARASCON, W. R. MCKINNON and L. H. GREENE, *Phys. Rev.* B35 (1987) 8843.
- P. E. SULEWSKI, A. J. SIEVERS, R. A. BUHRMAN, J. M. TARASCON, L. H. GREENE and W. A. CURTIN, *ibid.* B35 (1987) 8829.
- 4. Z. SCHLESINGER, R. L. GREENE, J. G. BEDNORZ and K. A. MULLER, *ibid.* **B35** (1987) 5334.

- 5. J. R. KIRTLEY, C. C. TSUI, SUNG I. PARK, C. C. CHI, J. ROZEN and M. W. SHAFER, *ibid.* **B35** (1987) 7216.
- 6. J. MORELAND, H. F. CLARK and H. C. KU, *Cryogenics* 27 (1987) 227.
- M. E. HAWLEY, K. E. GRAY, D. W. CAPONE II and D. G. HINKS, *Phys. Rev.* B35 (1987) 7224.
- J. R. KIRTLEY, R. T. COOLINS, Z. SCHLESINGER, W. J. GALLAGHER, R. L. SANDSTROM, T. R. DINGER and D. A. CHANCE, *ibid.* B35 (1987) 8846.
- 9. L. C. BOURNE, M. F. CROMMIE, A. ZETTL, HAN-CONRAD ZUR LOYE, S. W. WELLER, K. J. LEARY and D. E. MORRIS, *Phys. Rev. Lett.* **58** (1987) 2337.
- B. BATLOGG, R. J. CAVA, A. JAYARAMAN, R. B. VAN DOVER, G. A. KOUROUKLIS, S. SUNSHINE, D. W. MURPHY, L. W. RUPP, H. S. CHEN, A. WHITE, K. T. SHORT, A. M. MUJSCE and E. A. REITMAN, *ibid.* 58 (1987) 2333.
- T. FRELTOFT, J. P. REMEIKA, D. E. MONOTON, A. S. COOPER, J. E. FISCHER, D. HARSHMAN, G. SHIRANE, S. K. SINHA and D. VAKNIN, *Phys. Rev.* B36 (1987) 826.
- 12. D. C. JOHNSTON, J. P. STOKES, D. P. GOSHORM and J. T. LEWANDOWSKI, *ibid*. **B36** (1987) 4007.
- 13. J. M. TRANQUADA, D. E. COX, W. KUNNMANN, H. MOUDDEN, G. SHIRANE, M. SUENAGA, P. ZOLLIKER, D. VAKMIN, S. K. SINHA, M. S. ALVAREZ, A. J. JACOBSON and D. C. JOHNSTON, *Phys. Rev. Lett.* **60** (1988) 156.
- 14. M. STAVOLA, R. J. CAVA and E. A. REITMAN, *ibid.* 58 (1987) 1571.
- K. OHBAYASHI, N. OGITA, M. UDAGAWA, Y. AOKI, Y. MAENO and T. FUJITA, Jap. J. Appl. Phys. 26 (1987) L433.
- 16. K. MURALEEDHARAN and D. RAMBABU, *Phys. Rev.* **B36** (1987) 8918.
- K. K. SINGH, P. GANGULY and J. B. GOODENOUGH, J. Solid State Chem. 52 (1974) 254.

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